The Rooto Corporation

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Material Safety Data Sheet

Professional Drain Opener - 1069, 1071, 1079, 1084

Chem Trec Emergency 800-424-9300

Product Name:

Sulfuric Acid (H2SO4)

CAS Number:

7664-93-9

Synonyms:

Sulphuric Acid; Oil of Vitriol; Dipping Acid; Vitriol

Brown Oil; Hydrogen Sulfate

DOT Shipping Description:

Sulfuric Acid, 8, UN 1830, PGIL, RQ

II. Hazards Identification / Health Hazard Data

Hazard Communication Status:

Corrosive

Hazardous Components:

Sulfuric Acid 93.2%

Acute Health Effects:

<u>General</u> – Reacts with all body tissue and eyes to produce

severe burns and/or blindness.

Inhalation – Exposure to mists may cause irritation of nose,

throat and lungs, cough, headache, nausea, weakness.

Symptoms may be delayed.

<u>Ingestion</u> – Harmful or Fatal if swallowed. Product will severely burn the digestive tract. Damage may appear days

after exposure.

Chronic Health Effects:

<u>Inhalation</u> — Repeated exposure may cause severe shortness of breath, chronic bronchitis, chronic inflammation of the nose

and throat, corrosion/discoloration of teeth.

Skin Contact - Repeated exposure to mist or low

concentrated liquid may cause dermatitis.

Hazards Information:

TSCA Inventory Status:

Acute: Yes SARA Extremely Hazardous Substance: Yes

Chronic: Yes CERLCA Hazardous Substance: Yes Fire: No SARA Toxic Chemical: No Reactivity: Yes Carcinogens: None

Pressure: No

Health 3 3 3
Flammability 0 0
Reactivity 2

Ш. First Aid

Eye or Skin Contact:

Immediately flush with water for at least 15 minutes, while removing contaminated clothing. Get medical attention immediately. Apply cold, wet compresses; repeat flushing with cold water.

Remove from exposed area to fresh air. If not breathing, give Inhalation:

artificial respiration. If breathing is difficult, give oxygen.

Get medical attention immediately.

DO NOT induce vomiting. If conscious, give large quantities Ingestion:

of water. Get medical attention immediately.

Fire and Explosion Hazard Data

Non flammable. Flash Point:

Dry chemical or carbon dioxide. Water reacts violently with Extinguishing Media:

concentrated acid giving off heat.

Generates heat upon addition of water, with possible Special Fire Fighting Procedures:

splattering. Wear self-contained breathing apparatus and full protective clothing. Thoroughly decontaminate equipment

after use.

Will react with most metals, especially when dilute, to Unusual Fire and Explosion Hazard:

produce explosive Hydrogen gas. Will react with organic materials with evolution of heat and sulfur dioxide.

Storage and Handling

Employees must be properly trained to handle sulfuric acid. Handling:

Avoid breathing vapors or mist. Do not add water to contents. Do not smoke in a sulfuric acid storage area. Use nonsparking tools and vapor-proof type electrical fixtures.

Keep out of sun and away from heat sparks and flame. Keep Storage:

away from incompatible material. Storage tanks must be

ventilated. Do not add water to concentrated acid.

Accidental Release Measures/Waste Disposal

Keep people away. Prevent contact with skin. Wear full

protective equipment when handling sulfuric acid. Keep combustibles away from spilled material. Dike flow of spilled material with sand, gravel or whatever is available. Contact appropriate Federal, State or local officials. After obtaining approval from Federal, State or local officials, neutralize spill with soda ash or lime to a pH of 7. Prevent liquid from

entering sewers or waterways.

Follow Federal, State and local regulations for disposal. Make Disposal:

sure spilled material is neutralized to a pH of 7 and disposed of in an approved site. If not diluted and neutralized, this

waste would be a hazardous waste (RCRA

Hazardous Waste No. D002-Corrosive) as designated in 40 CFR 261. Reportable quantity is 1,000 pounds (based on

sulfuric acid content.)

VII. Applicable Control Measures

General ventilation must be provided. **Engineering Controls:**

Personal Protective Equipment:

Wear chemical safety goggles. Have eye bath and safety **Eve Protection:**

shower immediately available.

Wear full length face shield/chemical splash goggles Skin Protection:

combination, acid-proof gauntlet gloves, apron and boots, long-sleeved wool, acrylic or polyester clothing, acid-proof

suit and hood.

Respiratory Protection:

Use NIOSH approved equipment with full-face piece when

airborne exposure limits are exceeded.

Exposure Limits:

I mg/m3,8 Hr. TWA PEL (OSHA).

TLV (ACGIH): 1 mg/m3.8 Hr. TWA, A2 STEL 3 mg.m3, A2

VIII. Typical Physical Properties

None Odor: Colorless, oily liquid Appearance:

Solubility in water: Completely miscible 535°F (279°C) **Boiling Point:**

< 1 -31°F (-35° C) **Melting Point:** 1.835 Specific Gravity: Vapor Pressure: $< 0.6 \text{ mm Hg} (a) 100^{\circ} \text{F}$

Vapor Density: 3.4 < 0.3 mm Hg @ 77°F

Reactivity Data

Product is stable but reacts violently with water and organic **Chemical Stability:**

materials with evolution of heat.

Hazardous Polymerization:

Will not occur.

Vigorous reactions with water, alkaline solutions, metals, Incompatibility:

metal powder, carbides, chlorates, fuminates, nitrates, picrates,

combustible organic materials.

Hazardous gases evolve on contact with cyanides, sulfides, **Hazardous Decomposition Products:**

carbides. Hydrogen gas (highly flammable/explosive limits 4-75% by volume) is generated by the action of sulfuric acid on

Releases sulfur dioxide at extremely high temperatures. Decomposition:

Transportation Information

Hazardous Class: **Proper Shipping Name:** Sulfuric Acid Reportable Quantity: 1000 Pounds DOT Label: Corrosive Packing Group: **UN Number:** 1830

This information is for the specific material described only and may not be valid if the material is used in combination with any other materials or in any process. The user is responsible to determine the completeness of the information and suitability for the user's own particular use. To the knowledge and belief of GAC Chemical Corporation, the information is accurate and reliable as of the date indicated, but GAC Chemical Corporation makes no express or implied warranty of merchantability for the material or for the information. GAC Chemical Corporation makes no express or implied warranty of fitness for a purpose for the material or for the information.